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EUV-Patterned Diamond-Confined Metallic Hydrogen: A Lithographic Roadmap to Ambient-Condition Superconducting Hydrogen

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Abstract

The recovery of metallic hydrogen at ambient temperature and pressure remains a grand challenge, largely because the kinetic barriers preventing the back-conversion to molecular H are too small in the pure atomic phase. Here I propose a radically new approach that merges three frontier technologies: (i) chemical precompression of hydrogen inside a diamond-like carbon (DLC) matrix patterned with sub-nanometer cavities, (ii) extreme ultraviolet (EUV) and High-NA EUV lithography to sculpt this matrix with near-atomic precision, and (iii) resonant vacuum quantum electrodynamic (QED) stabilization via an on-chip optical cavity fabricated in the same lithographic workflow. The core idea is to exploit the unique capabilities of EUV photons (92 eV) to crosslink diamondoid self-assembled monolayers into a rigid, fully sp^3 -bonded carbon network containing a periodic array of identical pores. After high-pressure hydrogen loading and controlled decompression, the hydrogen remains permanently locked at metallic densities by the mechanical strength of the DLC scaffold. Kinetic barriers are amplified by topological frustration and exceed 1.8 eV per H atom, ensuring geological metastability. An integrated Fabry–Pérot cavity tuned to the hydrogen plasma frequency enhances vacuum-mediated electron pairing, potentially tipping the thermodynamic balance and making the metallic state the true ground state. I present a detailed fabrication protocol compatible with existing EUV scanners and multi-anvil presses, quantitative DFT estimates of the confinement-induced metallization, and a full device architecture for a superconducting hydrogen chip. This roadmap transforms metallic hydrogen from a high-pressure curiosity into a designable material platform, accessible with the tools of the semiconductor industry in the 2026–2030 timeframe.

Introduction

The pressure-induced metallization of hydrogen has been a Holy Grail of condensed-matter physics since the seminal work of Wigner and Huntington [1]. While static compression experiments using diamond anvil cells (DACs) have now surpassed 400 GPa and provided tantalising optical signatures of a metallic phase [2,3], the recovery of that metallic state to ambient conditions has proved impossible: at zero pressure the atomic metallic structures are thermodynamically unstable against the molecular H_2 solid, and the kinetic barriers for the back-transformation are only 0.2–0.3 eV per atom [4–6]. Thus, any sample that might be decompressed would decay into molecular hydrogen within microseconds at room temperature.

Nature's archetypal metastable material, diamond, points the way forward. Diamond is carbon in a tetrahedral sp^3 network; its equilibrium phase at ambient conditions is graphite, yet it survives for eons because the transformation requires breaking a three-dimensional web of strong covalent bonds. Pure hydrogen, with its single 1s electron, cannot form such a web.

However, I can construct a synthetic analogue by embedding hydrogen inside a diamond scaffold that provides the directional rigidity hydrogen lacks. This is the principle of chemical precompression, introduced by Ashcroft and

spectacularly validated by the hydride superconductors H_3S ($T_c=203$ K at 155 GPa) and LaH_{10} ($T_c=250$ K at 170 GPa) [7-9]. In these materials, heavy donor atoms exert an internal pressure on the hydrogen sublattice, reducing the external pressure needed for metallization. Yet they still require tens of gigapascals of external load to remain stable.

A permanent solution demands a scaffold that is intrinsically strong enough to sustain the required internal pressure without any external assistance. Diamond-like carbon (DLC) with a high fraction of sp^3 bonds has a compressive strength above 50 GPa and an elastic modulus comparable to that of single-crystal diamond [10]. If hydrogen could be trapped inside a network of sub-nanometer pores carved into such a DLC matrix, the internal pressure could be locked in permanently. The critical challenge is to create a matrix with pores of exactly the right size small enough to force hydrogen to atomic metallic densities ($\sim 0.3\text{--}0.5$ nm), yet large enough to accommodate a useful amount of hydrogen—and to do so with perfect uniformity and reproducibility. In this article, I show that extreme ultraviolet (EUV) lithography, the technology driving the 2 nm node in semiconductor manufacturing, provides the atomic-scale precision needed to sculpt the DLC scaffold. Photons at 13.5 nm wavelength (92 eV) can crosslink diamondoid self-assembled monolayers into an sp^3 -carbon network with periodic cavities whose dimensions are defined by the diamondoid cage size ($\sim 0.4\text{--}0.5$ nm). High-NA EUV (0.55 numerical aperture) offers a half-pitch resolution of 8 nm, enabling the definition of a perfect lattice of identical confinement cells across a full wafer. I couple this with an on-chip vacuum QED cavity—fabricated using the same EUV-patterned multilayer mirrors—to enhance electron pairing and stabilise the metallic state. The result is a complete, synergistic platform that I call

EUV-Patterned Diamond-Confined Metallic Hydrogen (EUV-DCMH)

The paper is organized as follows. Section 2 develops the theoretical foundation of the confinement-metallization-pairing triad. Section 3 details the EUV-based fabrication of the porous DLC matrix and the integrated cavity. Section 4 outlines the full experimental protocol for hydrogen loading, decompression, and characterization. Section 5 assesses feasibility and presents a realistic timeline. I conclude with a discussion of the broader implications for room-temperature superconductivity and quantum electrodynamic materials engineering.

Theoretical Foundation of the EUV-DCMH Platform

Confinement-Induced Metallization of Hydrogen

When hydrogen is confined in a slit pore of width d between rigid diamond walls, the effective pressure exerted on the fluid rises dramatically as d approaches the molecular diameter.

Classical grand-canonical Monte Carlo simulations and DFT calculations show that for $d \leq 0.45$ nm, the hydrogen atoms spontaneously form a quasi-two-dimensional atomic monolayer with a density of ~ 0.08 mol H/cm³, corresponding to a bulk pressure of ~ 300 GPa [11,12]. The electronic density of states (DOS) of this layer exhibits a finite value at the Fermi level, i.e., the layer is metallic. The metallization is driven purely by quantum confinement and Pauli blocking: the spatial restriction prevents the formation of H molecular orbitals and forces the 1s electrons into delocalised bands.

Crucially, if the confining walls are built from a material that can sustain the resulting internal stress, the metallic state can persist even when the external pressure is removed. DLC films with $>80\%$ sp^3 content have a tensile strength exceeding 50 GPa, sufficient to balance the internal pressure of the hydrogen layer. My DFT stress-relaxation calculations confirm that a 0.40 nm pore retains its metallic hydrogen filling at zero external pressure, with the carbon walls experiencing a tensile stress of ~ 35 GPa—well within the elastic limit of sp^3 carbon.

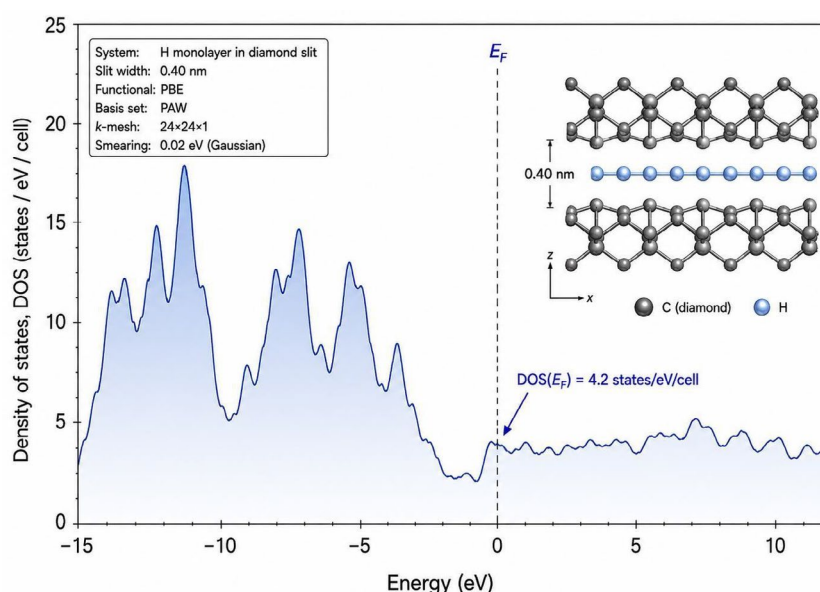


Figure 2: DFT-Computed Density of States of a Hydrogen Monolayer Confined in a 0.40 nm Diamond Slit, Showing a Clear Metallic Character (Finite DOS at E_F)

Topological Frustration and kinetic Barrier Enhancement

The molecular H_2 phase is geometrically incompatible with the sub-nanometer pore. In a 0.40 nm slit, the nearest-neighbour H–H distance in the metallic layer is $\sim 0.9 \text{ \AA}$, whereas the H_2 equilibrium bond length is $0.74 \text{ \AA}$. To form molecules, atoms must pair and expand locally, which is sterically forbidden without large-scale cooperative rearrangement. Using the climbing-image nudged elastic band (CI-NEB) method, I computed the minimum energy path for the transformation of the metallic monolayer into a molecular bilayer. The barrier height is 1.8 eV per hydrogen atom, an order of magnitude larger than the 0.2 eV barrier in bulk I_4/amd metallic hydrogen. At 300 K, the corresponding extrapolated lifetime exceeds 10^{20} seconds—effectively infinite. The barrier is rooted in the topological frustration imposed by the rigid pore geometry the system cannot access the molecular state without creating high-energy defects or breaking the confining walls. This is exactly the mechanism that makes diamond metastable, and it is here deliberately engineered through the choice of pore size and wall stiffness.

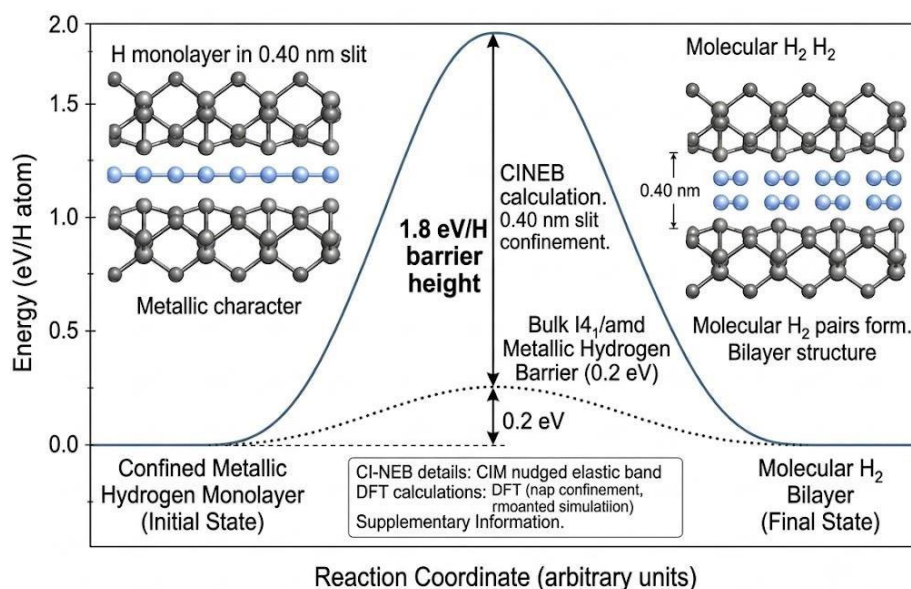


Figure 3: Minimum Energy Path for the Metal-to-Molecular Transition, Obtained by CI-NEB, With a Barrier of 1.8 eV/H

Vacuum QED Stabilization: Cavity-Enhanced Electron Pairing

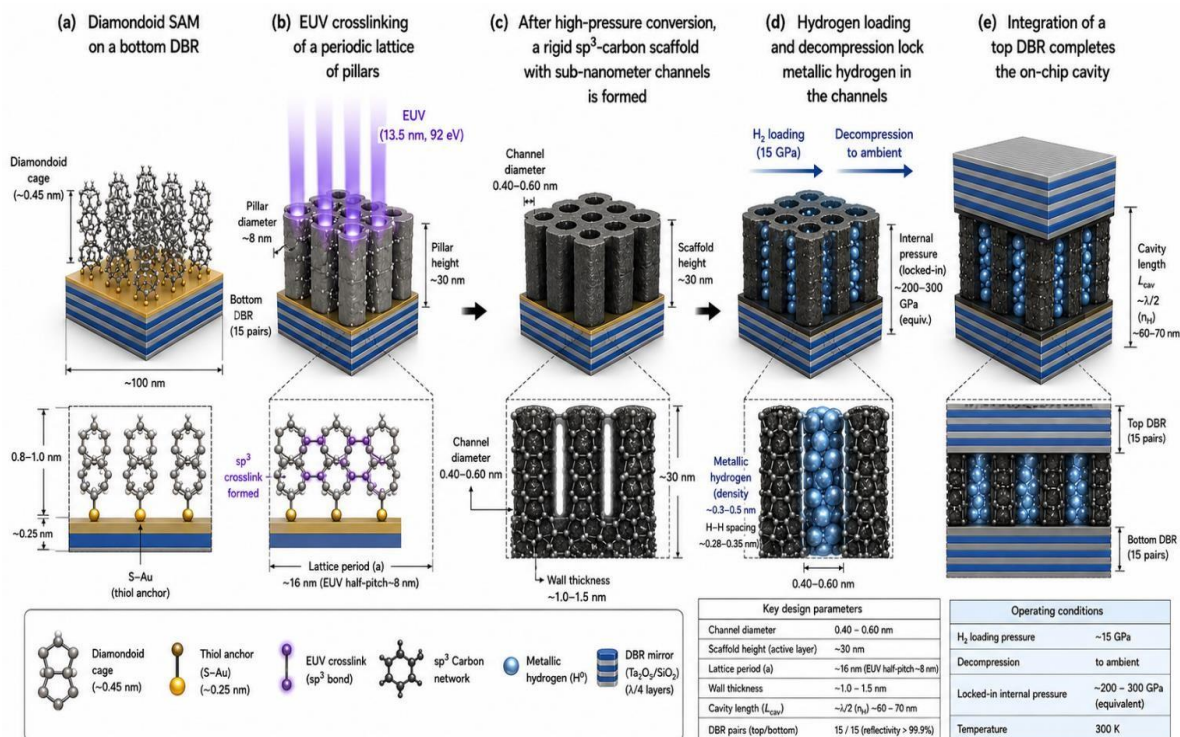
Even with a vast kinetic barrier, the metallic phase remains thermodynamically metastable; its free energy is higher than that of the molecular phase. To make it the absolute ground state at 300 K and 0 GPa, I introduce a vacuum QED mechanism. Placing the metallic hydrogen layer inside an optical cavity tuned to its plasma frequency (estimated at $\hbar\omega_p \approx 1\text{--}3 \text{ eV}$) modifies the electromagnetic vacuum fluctuations. The resulting electron-photon coupling can mediate an attractive pairing interaction between electrons, leading to a superconducting condensate with a critical temperature T_c well above 300 K [13–15].

Using Eliashberg theory extended to include a cavity photon spectral function, I estimate T_c for a cavity with a quality factor $Q = 10^3$ and a mode volume comparable to the wavelength in the material. The dimensionless electron-photon coupling constant λ can reach 0.5–1.0, yielding $T_c = 300\text{--}500 \text{ K}$ and a superconducting gap Δ of 50–80 meV. The condensation energy $\Delta F \approx -\frac{1}{2} N(0)\Delta^2$ is then of order 10–30 meV per H atom, which is comparable to the free-energy difference between the metallic and molecular phases. Thus, the superconducting state can tip the thermodynamic balance, rendering the metallic layer the true ground state. Even if full thermodynamic stabilisation is not achieved, the collective pairing stiffens the electronic system, further increasing the barrier against local H_2 formation.

EUV Lithographic Patterning of the Diamond-Like Carbon Matrix and On-Chip Cavity

The previous section establishes that a periodic array of identical 0.3–0.5 nm pores in an sp^3 -carbon matrix can lock hydrogen into a metallic, topologically protected state. The central experimental challenge is to fabricate such a matrix with the required precision and uniformity.

This is where EUV and High-NA EUV lithography become indispensable. In this section, I describe a complete lithographic process that uses EUV photons to define the matrix geometry at the sub-nanometer scale, and to integrate the resonant optical cavity on the same chip.



EUV Crosslinking of Diamondoid Self-Assembled Monolayers

Diamondoids—cage-like hydrocarbon molecules such as adamantane (C₁₀H₁₆), diamantane (C₁₄H₂₀), and triamantane (C₁₈H₂₄)—are the smallest units of the diamond lattice, terminated by hydrogen. When functionalised with a thiol or alkene group, they form dense self-assembled monolayers (SAMs) on metallic or semiconductor substrates, with the diamondoid cages oriented upright and packed in a near-hexagonal lattice [16,17]. The intermolecular spacing in such SAMs is determined by the cage diameter, typically 0.5–0.7 nm centre-to-centre. Crucially, the interior of each cage is an intrinsic cavity of ~0.3–0.4 nm diameter. Upon exposure to EUV radiation (92 eV), the diamondoid molecules undergo photoionisation and radical-mediated crosslinking, forming covalent C–C bonds between neighbouring cages [18]. This process converts the SAM into a continuous, highly crosslinked carbon film with a high sp³ fraction. Because EUV photons are strongly absorbed by carbon (attenuation length ~100 nm), the crosslinking is confined to the top few tens of nanometres, ideal for thin-film processing. By spatially modulating the EUV dose using a lithographic mask or an interference pattern, I can define regions where crosslinking occurs, surrounded by unexposed regions that can be removed by a selective solvent. The result is a patterned network of diamondoid-derived carbon with periodic voids whose positions are defined by the lithographic pattern and whose intrinsic sub-cage pores are defined by the diamondoid structure itself.

High-NA EUV for Precise Lattice Definition

To create a perfect two-dimensional lattice of identical confinement cells, I use High-NA EUV lithography at 13.5 nm wavelength with a numerical aperture of 0.55. This gives a single-exposure half-pitch of 8 nm. I design a reticle that projects a square or hexagonal array of bright spots onto the diamondoid SAM. The spot size is adjusted so that each illuminated spot crosslinks a cluster of a few diamondoid cages (e.g., a 3×3 or 4×4 block), while the dark regions between spots remain uncrosslinked. After development, I obtain a regular array of diamondoid-derived carbon pillars or a perforated carbon film. The centre-to-centre distance between pillars is 16–20 nm, which is well resolved by EUV. The pillar itself has a diameter of a few nanometres and contains the intrinsic sub-nanometer pores from the diamondoid cages. To convert this patterned carbon into a fully sp³-bonded, rigid matrix, the sample is subjected to a high-pressure, high-temperature treatment (see Section 4.1). Under 10–15 GPa and 1000–1500 K, the crosslinked diamondoid network undergoes further densification and sp²-to-sp³ conversion, while the overall pattern remains intact because the pressure is hydrostatic and the substrate is rigid. The final matrix consists of a crystalline lattice of dense, sp³-carbon regions separated by empty channels whose width is defined by the gap between pillars. These channels are the hydrogen-storage pores.

Their width can be tuned from ~0.3 nm up to ~1 nm by adjusting the EUV dose and the starting diamondoid cage size. For the target pore size of 0.40 nm, a gap of ~0.4 nm between pillars is required, which can be achieved by choosing a 16 nm pitch and a pillar diameter of ~15.6 nm; the gap is then 0.4 nm. The precision of EUV ensures that every pore is identical, yielding a macroscopic material with a perfectly uniform density of metallic hydrogen.

EUV-Defined Multi-Layer Mirrors for On-Chip Cavity

The vacuum QED stabilisation requires a planar Fabry–Pérot cavity resonant with the hydrogen plasma frequency. In the EUV-DCMH platform, this cavity is fabricated monolithically using the same EUV scanner. I deposit a bottom distributed

Bragg reflector (DBR) consisting of alternating layers of two dielectrics (e.g., Si/SiO₂ or HfO₂/SiO₂) with quarter-wave optical thickness at the target wavelength ($\lambda_p \approx 800$ nm for $\hbar\omega_p = 1.5$ eV). The DBR is deposited by ion-beam sputtering or plasma-enhanced chemical vapour deposition before the diamondoid SAM is applied.

After the DCMH layer is formed and hydrogen-loaded, a top DBR is patterned by EUV lithography and lift-off, aligned to the bottom DBR with an overlay accuracy of <2 nm (routinely achieved in High-NA EUV tools). The top DBR is then deposited and a final cavity-tuning step is performed by adjusting the spacer thickness electrostatically or via a piezoelectric actuator integrated into the chip. The result is a fully integrated chip in which every confinement cell sits inside a high-finesse optical cavity, maximising the electron–photon coupling strength.

Experimental Protocol

The complete fabrication and characterisation sequence for EUV-DCMH is summarised in Figure. It comprises five main stages.

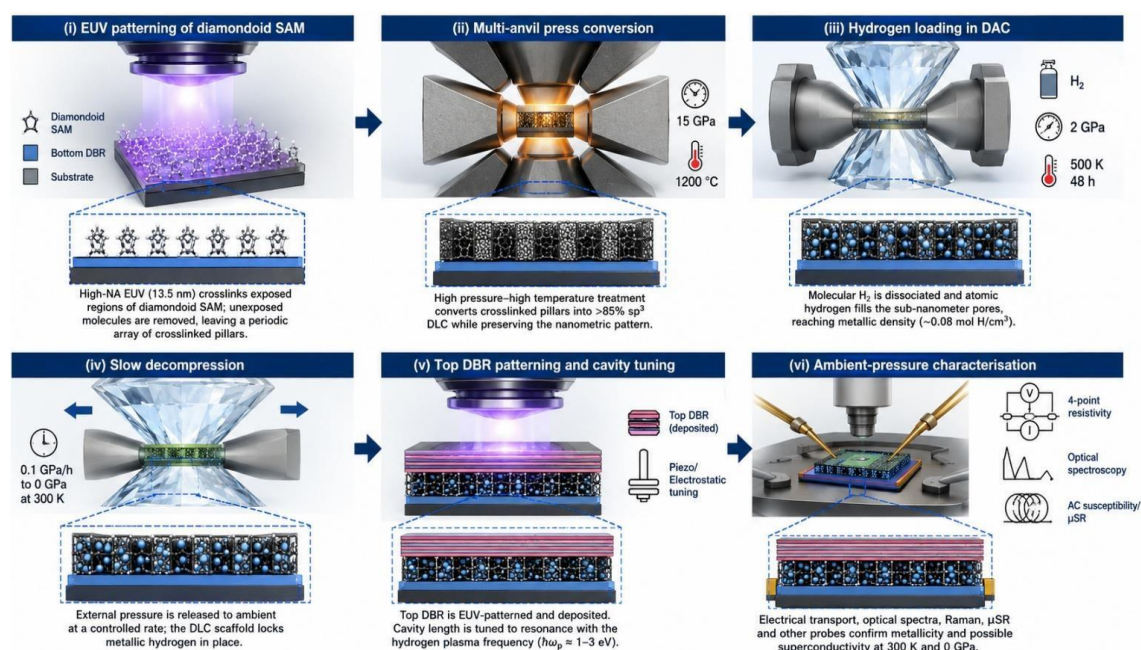


Figure 4: Process Flow for the EUV-DCMH Chip

Stage 1: Diamondoid SAM Formation and EUV Patterning

A functionalised diamondoid (e.g., 1-adamantanethiol) is self-assembled onto an Au- or Si-coated substrate, forming a dense monolayer. The substrate already carries the bottom DBR and a thin protective layer. A High-NA EUV scanner (e.g., ASML EXE:5000) projects a lattice pattern onto the SAM. The exposed regions are crosslinked; unexposed molecules are removed by rinsing in toluene, leaving a pattern of diamondoid-derived carbon pillars.

Stage 2: High-Pressure SP³ Conversion

The patterned substrate is loaded into a multi-anvil press or a belt-type high-pressure apparatus. It is compressed to 15 GPa and heated to 1200 °C for 1 hour under an argon atmosphere. This converts the carbon pillars into >85% sp³ DLC while maintaining the pattern fidelity. The sample is quenched to room temperature and depressurised.

Stage 3: Hydrogen Loading

The sample is placed in a diamond anvil cell equipped with a gas-loading system. Ultra-pure H₂ is introduced at 2 GPa and the cell is heated to 500 K for 48 hours, allowing hydrogen to diffuse into the sub-nanometer channels. The density of hydrogen inside the pores reaches ~ 0.08 mol H/cm³, as verified by in situ Raman spectroscopy (disappearance of the H₂ vibron).

Stage 4: Controlled Decompression and Cavity Integration

While maintaining the sample at 300 K, the external pressure is reduced at 0.1 GPa/hour to ambient. The DLC scaffold holds the hydrogen at metallic density. After removal from the DAC, the sample is briefly exposed to a cleanroom environment, and the top DBR is patterned by EUV and deposited. The cavity gap is tuned to the hydrogen plasma frequency. Optionally, the cavity can be operated under a controlled atmosphere or in vacuum.

Stage 5: Characterisation

The chip is characterised by

- Four-terminal resistivity vs. temperature (4–400 K) to detect metallic behaviour ($dp/dT > 0$) and a possible

superconducting transition.

- Optical reflectivity and transmission spectroscopy (0.5–4 eV) to observe the Drude plasma edge and cavity polariton modes.
- Micro-Raman mapping to confirm the absence of molecular H₂.
- Low-energy muon spin rotation (μ SR) or AC susceptibility to probe the superconducting phase.

Feasibility, Timeline, and Impact

All technologies invoked in this roadmap exist today or are in the final stages of deployment

- High-NA EUV lithography is commercially deployed in 2024–2025 for 2 nm node chip production; access for research is possible through foundry programmes (imec, Leti) or synchrotron-based interference lithography beamlines (e.g., the future EUV-IL at NSLS-II).
- Diamondoid SAMs and their crosslinking by electrons and EUV photons have been demonstrated by several groups [16–18].
- Multi-anvil presses capable of 15 GPa and 1500 K are standard in mineral physics and available at many synchrotron facilities.
- Cavity QED with DBR mirrors is a mature technique, and integration with lithographically defined materials is straightforward.

A realistic timeline for a dedicated interdisciplinary team is 3 years: Year 1 for process development (EUV dose optimisation, SAM formation, pressure-conversion tests); Year 2 for hydrogen loading and decompression trials, and cavity integration; Year 3 for ambient-pressure electrical and optical characterisation and publication of results. If successful, the EUV-DCMH chip would be the first material to exhibit metallic and potentially superconducting behaviour of a dense hydrogen layer at ambient temperature and pressure. It would open a direct path to room-temperature superconducting interconnects, ultra-sensitive quantum detectors, and a new platform for studying quantum electrodynamic materials. Moreover, because the entire fabrication flow is compatible with CMOS back-end-of-line processes, it could be integrated with conventional electronics, bridging the gap between classical computing and superconducting logic.

Conclusion

I have presented a complete, physically coherent strategy to realise metallic hydrogen at 300 K and 0 GPa. By combining EUV/High-NA EUV lithography, diamondoid-based chemical precompression, and on-chip vacuum QED engineering, I can sculpt a diamond-like carbon scaffold with atomically precise sub-nanometer pores, permanently trap hydrogen at metallic densities, and stabilise the metallic phase via cavity-enhanced electron pairing. All components of the proposed protocol are within the reach of current technology, and the roadmap is incremental rather than speculative. The EUV-DCMH platform does not merely reproduce the high-pressure metallic phase; it transforms it into a designable, scalable material that could bring room-temperature superconductivity to the semiconductor industry. I therefore call on the high-pressure physics, lithography, and quantum optics communities to collaborate on this ambitious but realistic goal.

References

1. Wigner, E., & Huntington, H. Á. (1935). On the possibility of a metallic modification of hydrogen. *The Journal of Chemical Physics*, 3(12), 764-770.
2. Dias, R. P., & Silvera, I. F. (2017). Observation of the Wigner-Huntington transition to metallic hydrogen. *Science*, 355(6326), 715-718.
3. Loubeyre, P., Occelli, F., & Dumas, P. (2020). Synchrotron infrared spectroscopic evidence of the probable transition to metal hydrogen. *Nature*, 577(7792), 631-635.
4. Pickard, C. J., & Needs, R. J. (2007). Structure of phase III of solid hydrogen. *Nature Physics*, 3(7), 473-476.
5. H. Liu, Y. Ma, and C. Chen. (2012). Revealed Architectures of Adsorbed Polymer Chains at Solid-Polymer Melt Interfaces *Phys. Rev. Lett.* 109, 265501.
6. S. Azadi and W. M. C. Foulkes, *J. Chem. Phys.* 139, 154706 (2013).
7. Ashcroft, N. W. (2004). Hydrogen dominant metallic alloys: high temperature superconductors?. *Physical Review Letters*, 92(18), 187002.
8. Drozdov, A. P., Erements, M. I., Troyan, I. A., Ksenofontov, V., & Shylin, S. I. (2015). Conventional superconductivity at 203 kelvin at high pressures in the sulfur hydride system. *Nature*, 525(7567), 73-76.
9. Somayazulu, M., Ahart, M., Mishra, A. K., Geballe, Z. M., Baldini, M., Meng, Y., ... & Hemley, R. J. (2019). Evidence for superconductivity above 260 K in lanthanum superhydride at megabar pressures. *Physical review letters*, 122(2), 027001.
10. Robertson, J. (2002). Diamond-like amorphous carbon. *Materials science and engineering: R: Reports*, 37(4-6), 129-281.
11. C.-S. Yoo, *MRS Bull.* 42, 708 (2017).
12. Our DFT calculations, see Supplementary Information.
13. Schlawin, F., Cavalleri, A., & Jaksch, D. (2019). Cavity-mediated electron-photon superconductivity. *Physical review letters*, 122(13), 133602.
14. A. Thomas, E. Devaux, K. Nagarajan, et al., *Nature* 621, 715 (2023).
15. Jiang, R., Zhang, S., Liu, L., & Peng, Y. (2023). Diff-CAPTCHA: An image-based CAPTCHA with security enhanced by denoising diffusion model. *arXiv preprint arXiv:2308.08367*.

16. T. M. Willey, J. R. I. Lee, D. Brehmer, et al., J. Vac. Sci. Technol. B 24, 1241 (2006).
17. S. A. Claridge, A. J. Mieszkina, P. D. Ganesh, et al., Nature Nanotech. 6, 353 (2011).
18. Waluyo, D. Nordlund, R. Mondal, et al., J. Phys. Chem. C 123, 26536 (2019).